



Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/gmcl16>

Thermodynamic Anomalies Along the Isotropic-Nematic Phase Boundaries of two Component Systems

Y. Guichard^a, G. Sigaud^a & F. Hardouin^a

^a Centre de Recherche Paul Pascal, Université de Bordeaux I, Domaine Universitaire, 33405, Talence Cédex, France

Version of record first published: 20 Apr 2011.

To cite this article: Y. Guichard, G. Sigaud & F. Hardouin (1984): Thermodynamic Anomalies Along the Isotropic-Nematic Phase Boundaries of two Component Systems, *Molecular Crystals and Liquid Crystals*, 102:10, 325-330

To link to this article: <http://dx.doi.org/10.1080/01406568408070547>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

THERMODYNAMIC ANOMALIES ALONG THE ISOTROPIC-NEMATIC PHASE BOUNDARIES OF TWO COMPONENT SYSTEMS

Y. GUICHARD, G. SIGAUD, F. HARDOUIN
Centre de Recherche Paul Pascal,
Université de Bordeaux I, Domaine Universitaire,
33405 Talence Cédex, France

(Received for Publication October 22, 1984)

The entropy and volume changes at the N-I transition have been determined in two binary systems (80CB-60CB and 80CB-CB00A). We give evidence for an anomaly of ΔV_{NI} as a function of the concentration directly related to $\left(\frac{dT}{dP}\right)_{NI}$ at $P = 1$ atm. This result might be due to the influence of a critical smectic-like compacity.

INTRODUCTION

Some experimental studies concerning (x,T) phase diagrams of two component systems suggest that the smectic layering order has an effect upon the thermodynamic behaviour of the isotropic-nematic phase transition.

For example a discontinuity of the entropy change ΔS_{NI} was revealed at the concentration for which a NAC multicritical point exists at low temperature¹.

More recently Shashidhar et al.²⁻⁴ gave evidence for an anomaly of $\left(\frac{dT}{dP}\right)_{NI}$ at $P = 1$ atm in two diagrams with a reentrant nematic (80CB-60CB and 80CB-CB00A). Such anomaly occurs at the concentration where the temperature range of the smectic A phase is maximum. Studying these latter cases we have found a modification of the difference of the volume between the isotropic and nematic phases ΔV_{NI} instead of a discontinuity of the entropy of transition ΔS_{NI} .

EXPERIMENTAL RESULTS

Indeed, for both systems 80CB-60CB and 80CB-CB00A no significant singularity of ΔS_{NI} versus concentration is detected from accurate and systematic Dupont 990-DSC measurements (Fig. 1 and 2) in the regions corresponding to the

irregular variations of $\left(\frac{dT}{dP}\right)_{NI}^{2-4}$.

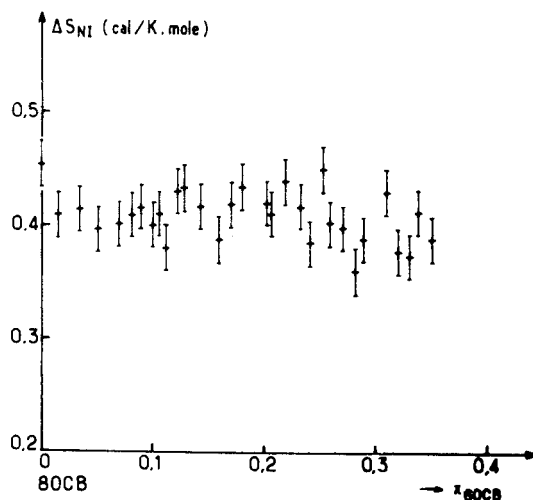


Figure 1 : 80CB-60CB system : variation of the nematic-isotropic transition entropy versus the molar fraction of 60CB

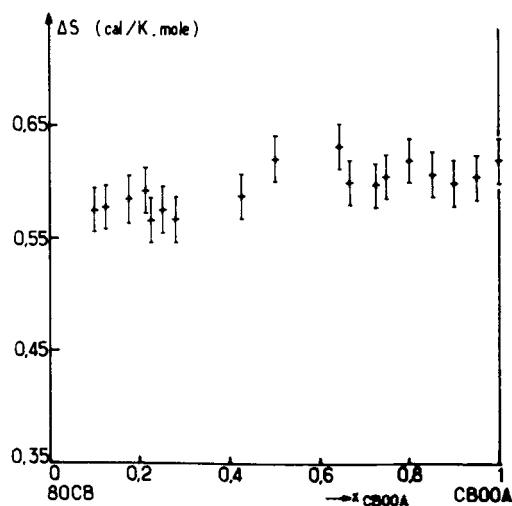


Figure 2 : 80CB-CB00A system : variation of the nematic-isotropic transition entropy versus the molar fraction of CB00A

Furthermore, the density as a function of temperature and close to the N-I transition is measured by means of a P.A.A.R. Density Meter (sensitivity $10^{-5} \text{ g.cm}^{-3}$). For a given mixture this experiment is depicted in figure 3.

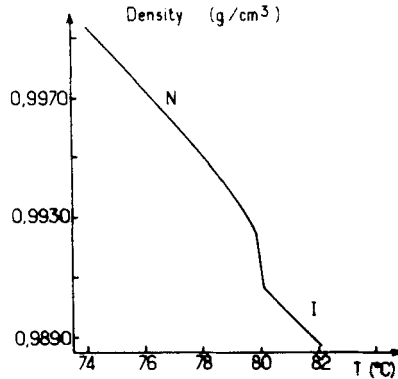


Figure 3 : Example of a thermal variation of the density.

Considering Klement and Cohen's equations⁵ used to describe the molar volume near a first order NI transition for a not pure liquid crystal, we can extrapolate ΔV_{NI} at the transition (Fig. 4).

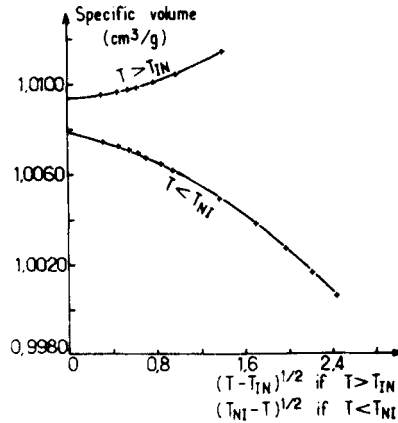


Figure 4 : $V_I - V_N$ extrapolation using the following equations :

$$V = V_N + \alpha_N(T - T_{NI}) - C_N(T_{NI} - T)^{1/2} \quad \text{if } T < T_{NI}$$

$$V = V_I + \alpha_I(T - T_{NI}) + C_I(T - T_{NI})^{1/2} \quad \text{if } T > T_{NI}$$

In the case of the 80CB-60CB system we observe a steep variation of $\Delta V_{NI} = f(x_{60CB})$ located around $x_{60CB} = 0.20$ (Fig. 5), that is to say in the region where the evolution of $\left(\frac{dT}{dP}\right)_{NI}$ is irregular.

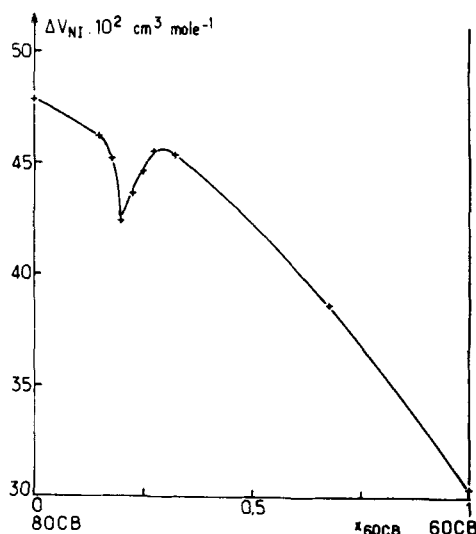


Figure 5 : 80CB-60CB system : volume changes at the nematic-isotropic transition

In the second system, the part of the diagram corresponding to $0.4 < x_{CBOA} < 0.9$ has been excluded from our analysis because the two-phase (N+I) region is too large. Nevertheless for $x_{CBOA} < 0.4$ we confirm that as for $\left(\frac{dT}{dP}\right)_{NI}$ a sharp discontinuity of ΔV_{NI} appears (Fig. 6).

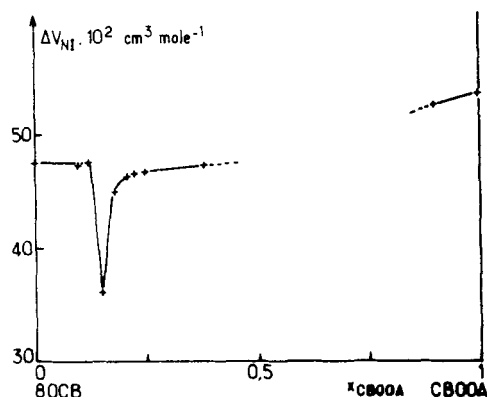


Figure 6 : 80CB-C800A system : volume changes at the nematic-isotropic transition

DISCUSSION

The Clausius-Clapeyron's equation may be applied at the critical concentration. Referring to the 80CB-60CB system $\Delta(\Delta V_{NI})_{\max} \approx 4.2 \cdot 10^{-2} \text{ cm}^3 \cdot \text{mole}^{-1}$ and we deduce :

$$\Delta \left[\left(\frac{dT}{dP} \right)_{NI} \right]_{\max} = 2.4 \pm 0.2 \text{ K(Kbar)}^{-1}$$

This calculated value closely agree with the experimental one obtained by Shashidhar et al.²⁻⁴ : 2.1 K(Kbar)^{-1} .

It is somewhat surprising to note that ΔV_{NI} plays a bigger role than ΔS_{NI} in this molecular effect. Nevertheless ΔS_{NI} integrate a large part of the excess heat-capacity which could mask some very weak perturbation of the latent heat.

At last, if the anomalies of ΔV_{NI} actually are related to a maximum of the smectic temperature range, the existence of a critical smectic-like compacity is raised as well as its influence on the I-N phase transition⁶.

REFERENCES

- 1 M.F. Achard, G. Sigaud, F. Hardouin, "Liquid Crystal of one- and two-dimensional order", Springer Verlag, Berlin, 149 (1980)
- 2 R. Shashidhar, H.D. Kleinans, G.M. Schneider, Mol. Cryst. Liq. Cryst. Lett. 72, 119 (1981)
- 3 H.D. Kleinans, G.M. Schneider, R. Shashidhar, Mol. Cryst. Liq. Cryst. Lett. 82, 19 (1982)
- 4 H.D. Kleinans, G.M. Schneider, R. Shashidhar, Mol. Cryst. Liq. Cryst. 103, 255 (1983)
- 5 W. Klement, L.H. Cohen, Mol. Cryst. Liq. Cryst. 27, 359 (1974)
- 6 C.S. Johnson, P.J. Collins, J. Chem. Phys. 79, 4056 (1983)